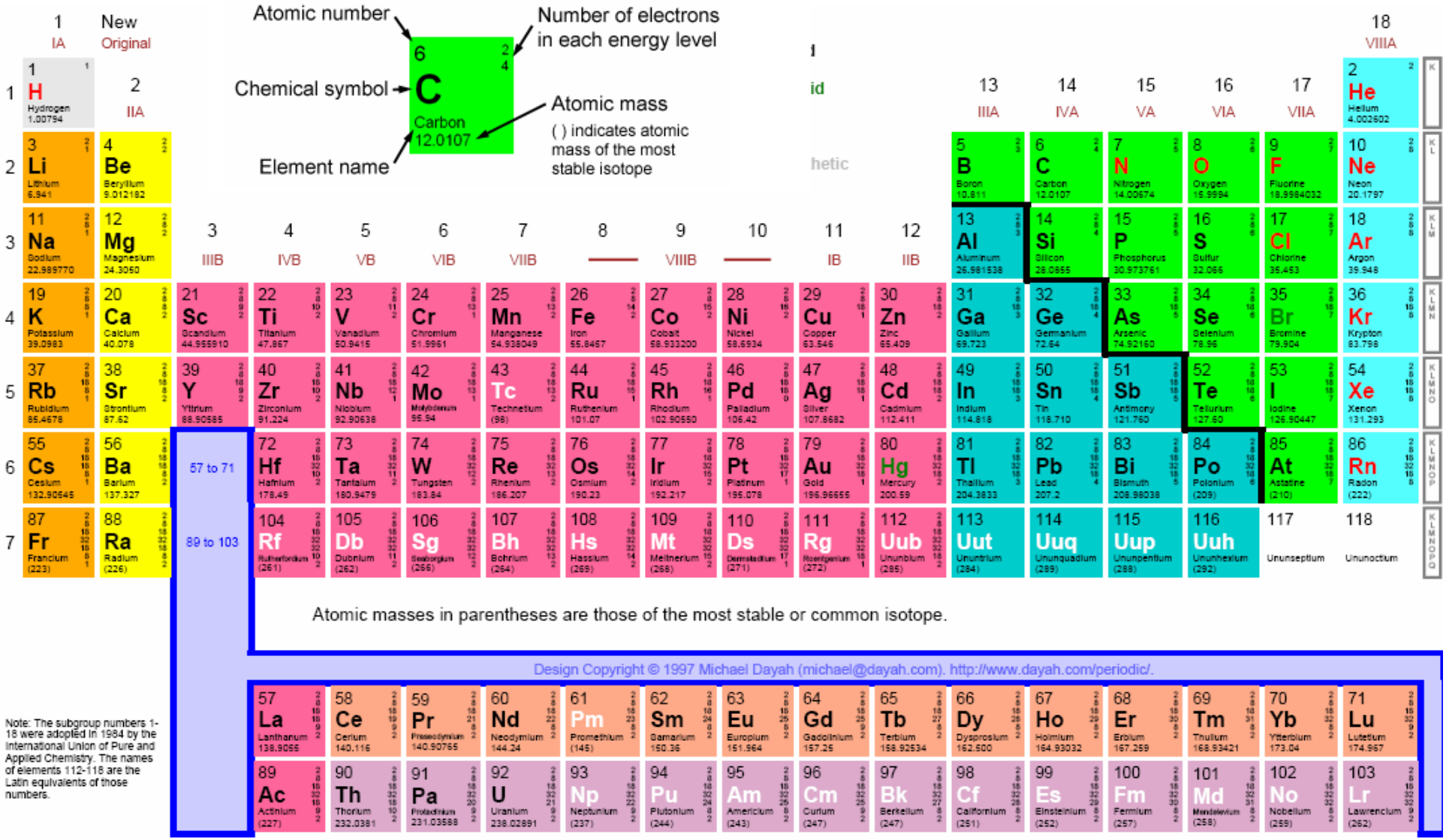
附錄三：元素週期表


Atomic number
Number of electrons in each energy level
Chemical symbol
Atomic mass
( ) indicates atomic mass of the most stable isotope
Element name
6
C
Carbon
12.0107
1
New
Original
IA
IIA
IIIB
IVB
VB
VIB
VIIB
VIIIB
IB
IIB
IIIA
IVA
VA
VIA
VIIA
VIIIA
57 to 71
89 to 103
Atomic masses in parentheses are those of the most stable or common isotope.
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Note: The subgroup numbers 1-18 were adopted in 1984 by the International Union of Pure and Applied Chemistry. The names of elements 112-118 are the Latin equivalents of those numbers.