

東吳大學 105 學年度轉學生(含進修學士班轉學生)招生考試試題

第 1 頁，共 2 頁

系級	微生物學系二年級	考試時間	100 分鐘
科目	普通化學	本科總分	100 分

- What is the atomic number of Phosphorous (P)? (5%)  
(A) 31, (B) 15, (C) 209, (D) 84
- What is the name of the element with the symbol Fe? (5%)  
(A) Selenium, (B) Fermium, (C) Iron, (D) Francium
- Indicate the oxidation state of the underlined elements in (A) P<sub>4</sub> (B) FeO<sub>4</sub><sup>2-</sup> (10%)
- Which of the following sets of quantum numbers is not matched with the orbital designation. (5%)  
(A)  $n = 1, \ell = 0$ : a 1s orbital  
(B)  $n = 2, \ell = 0$ : a 2p orbital  
(C)  $n = 3, \ell = 0$ : a 3s orbital  
(D)  $n = 3, \ell = 1$ : a 3p orbital  
(E)  $n = 3, \ell = 2$ : a 3d orbital
- Which orbital-filling diagram represents the anomalous ground state of chromium? (5%)  
(A)  

[Ar]	$\uparrow\downarrow$	$\uparrow$	$\uparrow$	$\uparrow$	$\uparrow$	—
	4s			3d		

  
(B)  

[Ar]	$\uparrow\uparrow$	$\uparrow$	$\uparrow$	$\uparrow$	$\uparrow$	—
	4s			3d		

  
(C)  

[Ar]	$\uparrow$	$\uparrow$	$\uparrow$	$\uparrow$	$\uparrow$	—
	4s			3d		

  
(D)  

[Ar]	$\uparrow\downarrow$	$\uparrow\downarrow$	$\uparrow$	$\uparrow$	—	—
	4s			3d		
- What is the ground-state electron configuration of Co? (5%)  
(A) [Ar]3d<sup>9</sup> (B) [Ar]4s<sup>1</sup>3d<sup>8</sup> (C) [Ar]4s<sup>2</sup>3d<sup>7</sup> (D) [Ar]4s<sup>2</sup>4p<sup>6</sup>4d<sup>1</sup>
- Name these compounds (a) and (b), and write formulas for the compounds (c) and (d): (20%)  
(a) SrO (b) ZnS (c) silicon tetrafluoride (d) 3-pentanol
- To what volume should you dilute 0.200 L of a 15.0 M NaOH solution to obtain a 3.00 M NaOH solution? (10%)
- Titanium metal can be obtained from its oxide according to the balanced equation: 背面尚有試題  

$$\text{TiO}_2(s) + 2 \text{C}(s) \rightarrow \text{Ti}(s) + 2 \text{CO}(g)$$

When 28.6 kg of C reacts with 88.2 kg of TiO<sub>2</sub>, 42.8 kg of Ti is produced. Find the limiting reactant, theoretical yield (in kg), and percent yield. (M.W. C = 12.01, Ti = 47.87, O = 16.0) (15%)

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第 2 頁，共 2 頁

系級	微生物學系二年級	考試時間	100 分鐘
科目	普通化學	本科總分	100 分

For the reaction:  $A + B \rightarrow C + D$

Experiment	[A], M	[B], M	Initial rate, $M s^{-1}$
1	0.185	0.133	$3.35 \times 10^{-4}$
2	0.185	0.266	$1.35 \times 10^{-3}$
3	0.370	0.133	$6.75 \times 10^{-4}$
4	0.370	0.266	$2.70 \times 10^{-3}$

10. What is the order of reaction with respect to A? (5 %)

(A) 1 (B) 2 (C) 3 (D) 4

11. What is the order of reaction with respect to B? (5 %)

(A) 1 (B) 2 (C) 3 (D) 4

12. What is the overall reaction order? (5 %)

(A) 1 (B) 2 (C) 3 (D) 4

13. What is the value of the rate constant, k? (5 %)

(A) 0.77 (B) 0.07 (C) 0.55 (D) 0.10